The following note is based on the Coursera Mooc **Density Functional Theory** by **École Polytechnique** which is available at https://www.coursera.org/learn/density-functional-theory

Week 1: From many-body problem to DFT

The many-body problem: We start by writing the many-body Hamiltonian:

$$\hat{H} = -rac{h^2}{2m_e} \sum_i
abla_i^2 - \sum_{i,I} rac{Z_I e^2}{|r_i - R_I|} + rac{1}{2} \sum_{i
eq j} rac{e^2}{|r_i - r_j|} - \sum_I rac{h^2}{2M_I}
abla_I^2 + rac{1}{2} \sum_{I
eq J} rac{Z_I Z_J e^2}{|R_I - R_J|}$$

Here, i,j range over the number of electrons, and I,J range over the number of nuclei. Since the mass of the neclei is much larger than of electron, we can neglect the kinetic energy term due to nuclei. Also, the Coulomb interaction term between nuclei is just a constant since there movement in a system can be ignored. So, what ultimately remains is:

$$\hat{H} = -rac{h^2}{2m_e} \sum_i
abla_i^2 - \sum_{i,I} rac{Z_I e^2}{|r_i - R_I|} + rac{1}{2} \sum_{i
eq j} rac{e^2}{|r_i - r_j|}$$

Now there might be some other potential coming outside from the system. We call all such potential in addition with the potential coming from the nuclei as the external potential and we write:

$$\hat{H} = -rac{h^2}{2m_e} \sum_i
abla_i^2 - \sum_i U_{ext}(r_i) + rac{1}{2} \sum_{i
eq j} rac{e^2}{|r_i - r_j|}$$

However, even this simplified equation is impossible to tackle exactly.

Observables and Functional:

What we want from a system are the observables. In QM, they are the expectation value. Let O be an observable. Then, in μVT ensemble:

$$\langle O=\langle \hat{O}
angle =rac{1}{Z}\sum lpha e^{-eta(E_lpha-\mu N_lpha)}raket{\psi_lpha|\hat{O}|\psi_lpha}$$

That means the observale can be expressed as a functional of the wavefunction. However, the wavefunction is too complicated to calculate. That's why one way to approach is to approximate the wavefunction. Some of the such approaches are: Hartree-Fock, Quantum Monte Carlo etc.

However, we could rewrite the functional in terms of completely something else such as instead of ψ , we can use U_{ext} . So, instead of $O[\psi]$, a functional of wavefunction, we are looking for

 $\widetilde{O}[U_{ext}]$, a functional of the external potential. But now, the functional would be too difficult to evaluate.

 $O[\psi]$: Simple functional of complicated function $\widetilde{O}[U_{ext}]$: Complicated functional of simple function

That's why we are looking for another functional. In this course, we will see that the functional will be of electron density, $\rho(r)$, i.e., we are looking for:

$$\mathcal{O}[\rho(r)]$$
: The goal

Also, we want different functionals of this same function for all observables that would make our life easier. Observables can include but not limited to equlibrium bond angle & length, ground sate properties, etc.

We skip the functional derivative part.

Hohenberg-Kohn (HK) theorem: We know the statements of two HK theorems from the lecture note of *Lee (Chapter 5)*. Let us give here a proof scheme for the first theorem.

Let two external potential differ by more than a constant, i.e., $U_{ext}^{(1)} \neq U_{ext}^{(2)} + {
m const.}$ and they gave rise to same ground state density. Therefore,

$$egin{aligned} H_1 |\psi_g
angle &= E_1 |\psi_g
angle \ H_2 |\psi_g
angle &= E_2 |\psi_g
angle \ &\Rightarrow \left[U_{ext}^{(1)} - U_{ext}^{(2)}
ight] |\psi_g
angle &= (E_1 - E_2) |\psi_g
angle \ &\Rightarrow U_{ext}^{(1)} = U_{ext}^{(2)} + ext{const.} \end{aligned}$$

which violates our assumption. Therefore, there cannot be two different external potential leading to the same ground state electron density.

An important remark: HK theorem only extablishes the uniqueness of the external potential and the electron density. It says nothing about the existence of such functional.